

Limnia Plasma Hydride Array System Engineering (PHASE Technology)

Presented by: LIMNIA under NDA

Overview

- \bullet Project outline
- \bullet **Background**
- \bullet New technology array rig
- \bullet 1st generation metal-hydrides
- \bullet 2nd generation metal-hydrides
- \bullet Metal doped glass structures
- \bullet Photo-catalysis
- \bullet **Conclusions**

Limnia "PHASE" - Productization Outline

- • Develop, engineer and generate highly efficient array system engineering for Limnia (PHASE) technologies.
- • Produce the ideal generation of novel materials for the storage of hydrogen using the above technologies.
- • Ongoing protection by patent/copyright of technology and license of the new systems to external companies for production and distribution of new materials.

Background

What is ion gas?

- •It is ionised gas containing free electrons and ions.
- • In Limnia technology, array devices are used to generate the conductive arena. Direct current (DC) is generally used and the arc is stabilized by using magnetic fields, gas injection & RF generators.
- • Array processes are characterized by short response time, good process control, high yield, high and consistent product quality, and high production per unit volume.
- • The technology meets the demands for more environmentally friendly processes.

Why use this approach?

- • Gas-phase synthesis gives the best control over purity, size, shape and crystallinity. Very high chemical purity is practical with gas-phase synthesis whereas it is problematic with many of the other methods. Array heating provides a clean, directional, controllable, high intensity, localised source of contaminant free heat. Other industries have used plasma but it is little-developed for hydrogen.
- \bullet In the array zone (operating at up to 20k degrees Kelvin) a range of novel materials can be produced. Limnia's approach concentrates on spherical nanometric powders with diameters in the range 15 - 300 nm, with exceptionally high specific surface area/mass ratios, up to >200m 2 g ⁻¹ .

Advantages over conventional routes

- •No contaminants.
- •Work under an inert atmosphere (H $_{2}$, Ar, N $_{2}$ etc.)
- •Spherical particles can be formed from solids and facetted particles.
- •Fast process cycles, often in minutes, with no need for preheating time.
- •Rapid start up and attainment of steady state.
- •Compact equipment and small footprint of array rigs.
- •Single rig can make range of products.
- • Inter-changeablility of services allowing range of rigs to share service infrastructure.

New Technology (PHASE)

PHASE I Fab Rig

- • In order to improve current new alloys and to reduce costs, a fab rig is being constructed in order to optimize alloy candidates.
- \bullet Conductive array gun to be used in the process is rated at 100kW. (Approx size = 10cm x 6cm x 2cm) i.e. it is very small/compact.
- • 100kW array core torch can produce a 50g batch of material in 30 minutes.
- • The diversity of the process allows the screening and production (very fast) of a wide range of novel/new alloys, as media.
- \bullet A novel architecture, never before deployed by competitors, with proprietary system subsets unique and exclusive to Limnia.

PHASE I Fab Rig

"*Sanitized*" Simple Boxchart Schematic diagram of PHASE I Fab rig.

PHASE I Processing Components

- •Feed single materials and mixtures and volatile liquids.
- •Material feed rates and method of presentation to the array plume.
- • Conductive array torch/electrode type, power input and distribution, spatial configuration and gas flow rate.
- \bullet Quench gas identity, rates and timing in the reaction
- •Temperature variation and oxidation/reduction processes
- •Collection process for end product (directly into Limnia cassettes).
- • These variables allow Limnia's array system to be adapted to make a range of materials.

PHASE I Generation Metal Hydrides

PHASE I M-H materials

- \bullet Metal powders (Al, Mg, Sn, Ti etc) and liquids (Fe(CO) $_5$) of known high weight percentage ratios with hydrogen are injected into the array.
- \bullet Works under inert atmosphere (N $_{2}$, Ar, H $_{2}$ etc)
- \bullet Metals are vaporised into ions, electrons and small clusters.

Schematic of vaporisation and condensation

• On cooling (cryosystem or collision gas), the metal particles undergo condensation/coalescence forming nano-sized crystals (few nm in size).

PHASE I M-H materials

- \bullet Target metal species are highly reactive and will "pick-up" collision gas molecules (H $_{\rm 2}$ preferable), thus forming M-H $\,$ materials (H $_{\rm 2}$ also acts to stabilise the metal ready for collection).
- Stoichiometry (tandem feed) is easily controlled, leading to a vast diversity of doped, mixed metal (alloy) metal hydride systems.
- •The process of isolation of the materials is very fast (i.e. new materials for characterisation and kinetic testing may be produced every few hours).
- Characterisation of the materials and further testing will be carried out using standard techniques, such as XRD, SEM etc.

PHASE II Generation Metal Hydrides

Supported M-H materials

- \bullet The supported metal catalysts used widely in technology consist of aggregates (sometimes called clusters of crystallites or particles) of metal of various sizes and shapes dispersed on a support.
- Aggregates (metal catalysts) smaller than about a few nm may be the most important catalytically (hence the use of plasma spraying) because they have a large fraction of metal exposed at the surface and therefore accessible to reactants (hydrogen).
- \bullet These very smaller clusters are of most interest structurally, since they resemble molecular species more than bulk metals.

Supported M-H materials

- • Direct spraying within the array onto alumina/silica supports.
- \bullet Supports introduced with collision/buffer gas.
- \bullet Fine layer (few nm of metal) produced on the surface.
- \bullet Metal Alloys (mixed metal catalysts).

SEM of Fly Ash (one of many intended supports).

Supported M-H materials

- \bullet One of the main problems encountered in hydrogen storage, is the loss of hydrogen to the metal subsurface. i.e. it requires more energy to desorb it from the metal interior than the surface (higher temperatures).
- \bullet Low temperature adsoprtion and desorption.
- \bullet High weight percentage of material to hydrogen ratio.

Schematic of a supported catalyst

M-H materials

- •**Weight Percentage (%)**
- • For example the 3d metals and main group elements (Al, Mg, Ti, Si and Zn) are the primary focus of this Limnia compound generation.
- • Weight percentages in the order of 7-10 % will be achieved by this method.
- \bullet However, 20 - 40% by weight is potentially possible per our tests & analysis. 2 g/cc to 7g/cc are possible with this solution.
- •**Storage efficiency (%)**
- • The efficiency of these materials will be greater than 90%. The materials are perfect in purity, size, composition, shape and crystallinity.
- • No loss of Hydrogen to the system. Therefore should see 100% adsorption/desorption.

Phase III: Metal Doped Glass Microstructures

Metal doped glass spheres

- • Coating material + metal will be introduced into the conductive array system.
- •Coating of the metallic particles (encapsulation).
- • Silica, alumina, diamond, boron nitride etc. are deemed possible candidates for coating the metallic particles.
- •Metals/alloys - e.g. Lithium (+10 wt% hydrogen)

Metal powder + support plasma entity between the encapsulation

Metal doped glass spheres

- The glassy coating encapsulating the metal particles upon heating and pressure will allow hydrogen to diffuse into the micrsphere on/into the metal.
- The metal will act as a storage medium + giving vital stability (from cracking) to the glass spheres.
- The metal will be protected (inert atmosphere) from gas molecules (e.g. O_2) and $\mathrm{H}_2\mathrm{O}.$

PHASE IV Compounds

PHASE IV Metal oxo semiconductors

- Metal powders e.g. (indium, tantalium, nickel etc.) will be introduced into the array in an oxygenated atmosphere (Oxygen Carrier Gas).
- A wide variety of novel mixed metal catalysts can be generated (cf. PHASE I).
- Powders may be supported as in PHASE II. Providing higher surface area of the Catalyst.
- All PHASE I, II, III properties apply to PHASE IV.

Conclusions

- The Limnia PHASE project provides an excellent opportunity for the diverse development and production of Limnia's novel technologies and materials.
- The PHASE system will provide inexpensive, environmentally friendly systems and materials.
- All the above PHASE projects are currently Patent Pending under Limnia filings.

Thank you

- •Limnia
- \bullet 601 Van Ness Avenue, Suite E3613
- \bullet San Francisco, CA 94102
- •510.868.2862
- •www.Limnia.com