

# Hydrogen Storage in Graphite Nanofibers and the Spillover Mechanism

A Study Carried Out in the DOE Center of Excellence  
on Carbon-based Hydrogen Storage Materials

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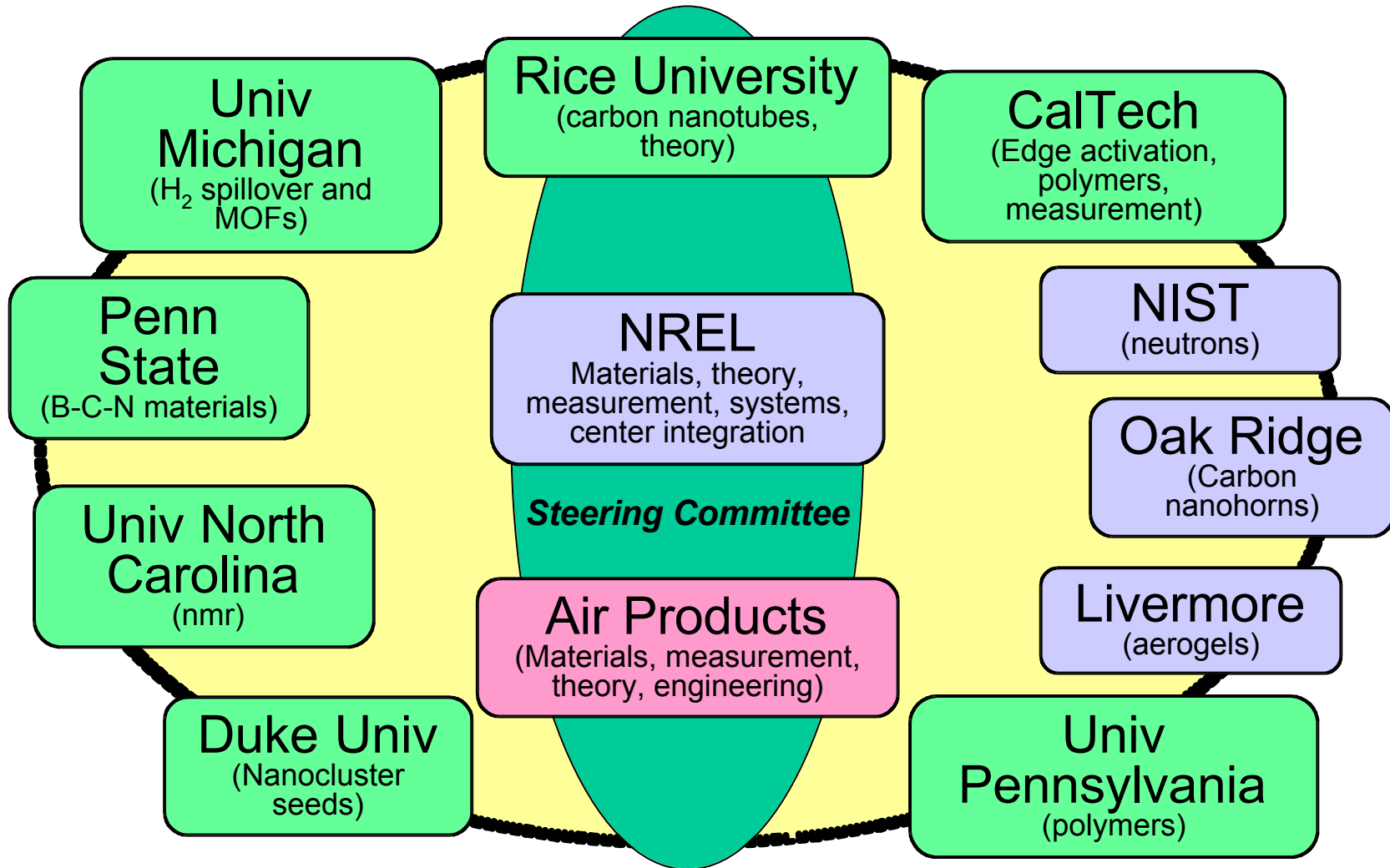
Department of Chemical Engineering

23-24 May 2005

# CbHS

## Center of Excellence Partners

*9 university projects (at 7 universities), 4 government labs, 1 industrial partner*



# Overview

## Timeline

- Project start date: FY05
- Project end date: FY09
- New Start

## Budget

- Expected Total Funding
  - DOE share: \$939,356
  - Contractor share: \$280,000
- Funding for FY05: \$170,000

## Barriers

- General
  - Weight & Volume
  - Efficiency
- Reversible Solid-State Material
  - Hydrogen Capacity & Reversibility
  - Lack of Understanding of H Physi- & Chemisorption

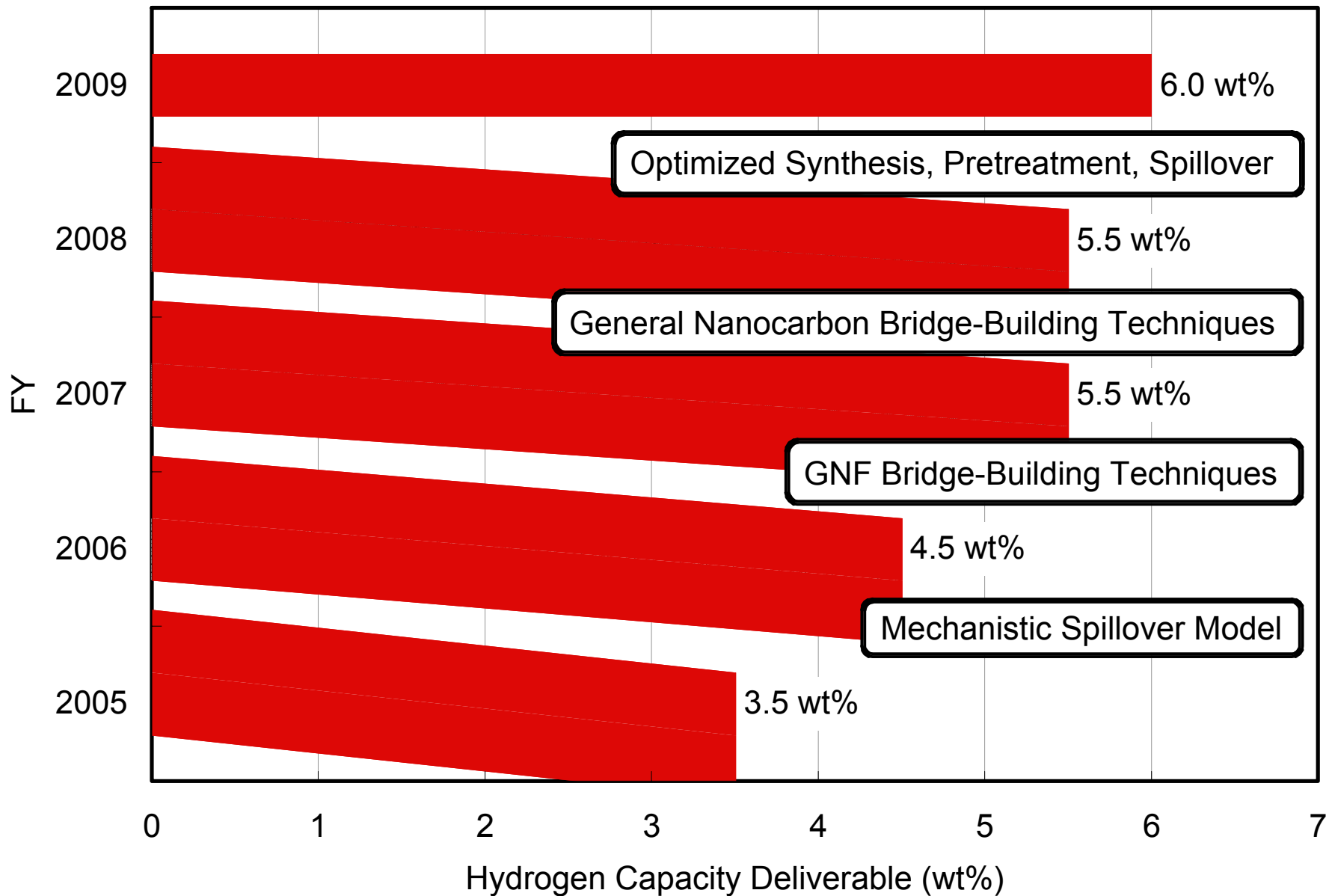
## Partners

- Sample/adsorbed H Measurements and Characterization
  - NREL, NIST

# Project Objectives

- To develop Graphite Nanofiber (GNF) based hydrogen storage materials with capacities in excess of 6 wt%
  - To Optimize GNF Synthesis Catalyst & Pretreatment Conditions for Hydrogen Storage
  - To Develop Bridge-Building Techniques for Spillover to Enhance Hydrogen Storage
  - To Obtain a mechanistic understanding for hydrogen spillover in nanostructured carbon materials

# Project Milestones



# Approach

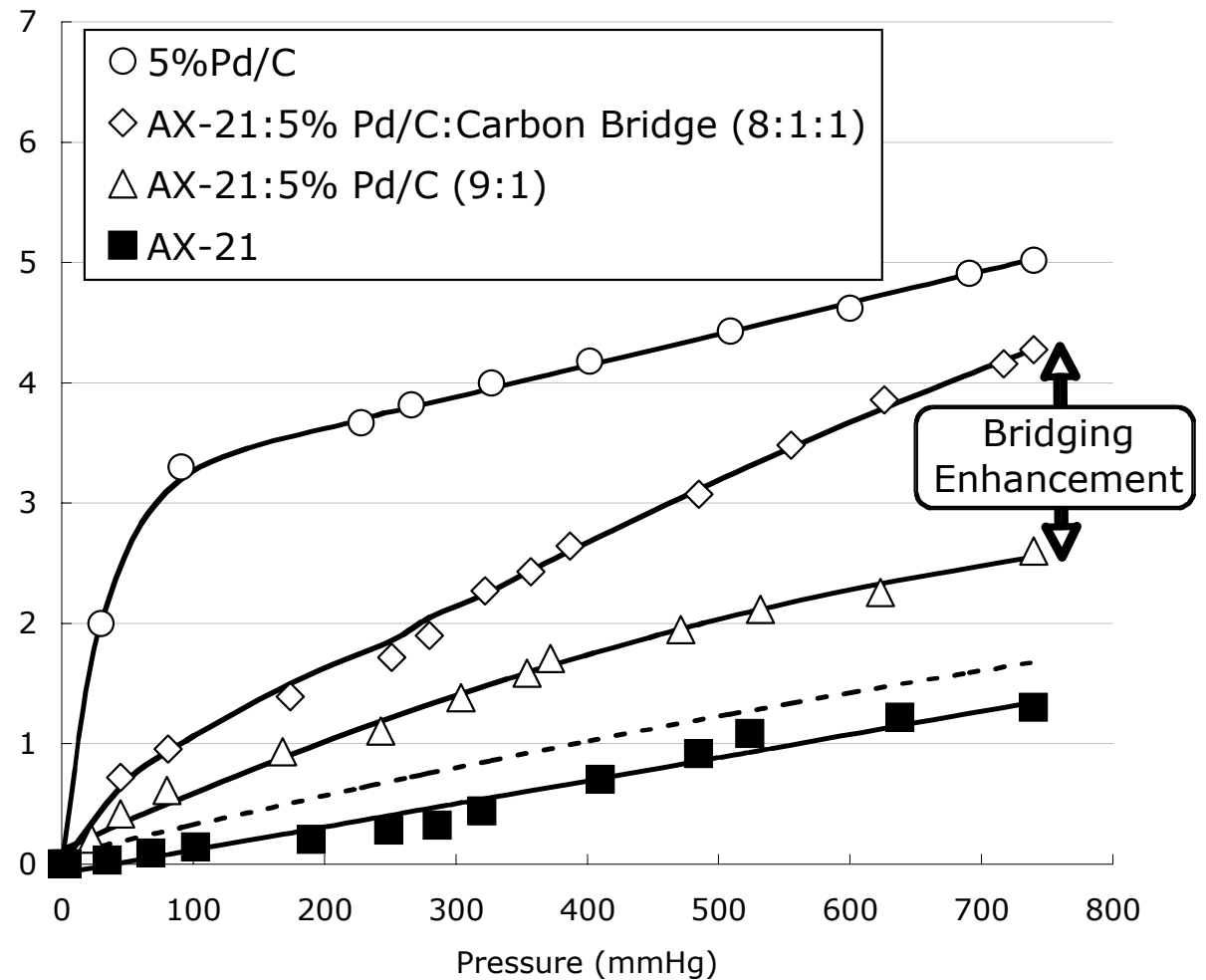
- Development of GNF material with  $> 6 \text{ wt}\%$  reversible hydrogen capacity
  - Systematic studies of GNF synthesis catalysts (metal alloys) & pretreatment conditions
  - Production of composite materials containing catalysts to dissociate hydrogen
  - Enhancement & optimization of spillover mechanism through modeling & bridge-building techniques

# Technical Accomplishments

- Development of Calibrated High Pressure Apparatus & Test Protocol
- Demonstration of Bridge-Building Technique to Enhance Spillover
- Screening of Two GNF-Metal Hydride Composites for Spillover Enhancement
- Identification of Carbon Composite Demonstrating Reversible Hydrogen Storage Capacity of 1.8 wt% at 298 K & 10 MPa

# Low Pressure Spillover Enhancement

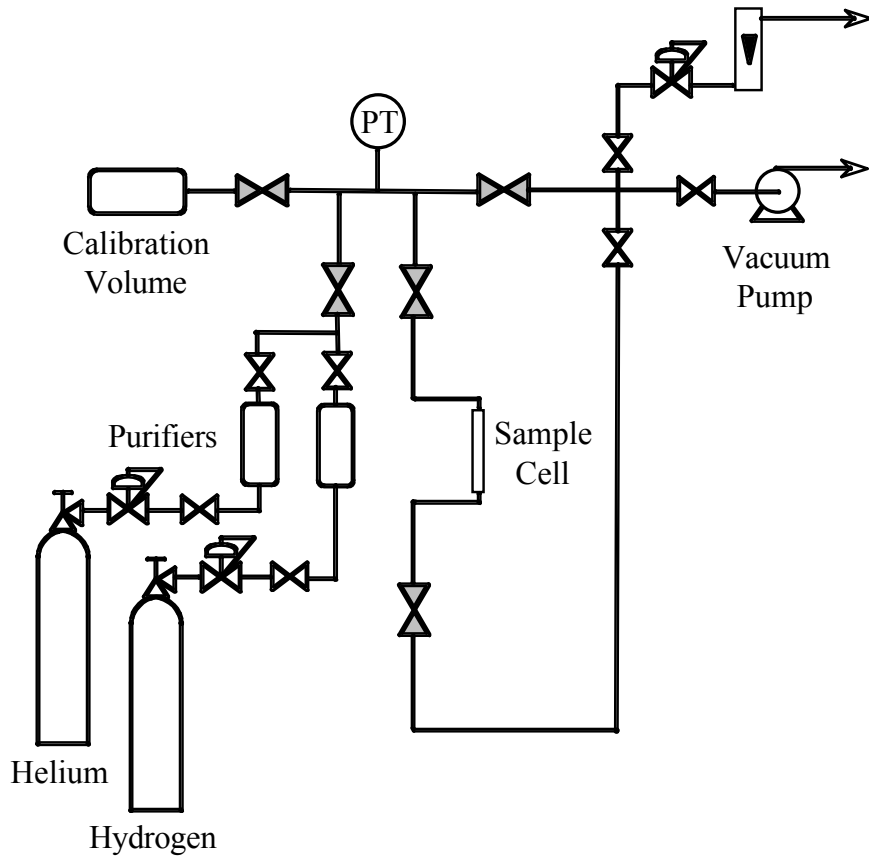
- AX-21 Receptor
- 5 wt% Pd/C Catalyst
- Carbon 'bridge' formed by carbonization of precursor (e.g., glucose)
- Adsorption at 298 K
- < 4% adsorbed volume at 0.1 MPa due to  $\text{PdH}_{0.6}$
- Adsorption capacity tripled at 0.1 MPa (only doubled without bridges)



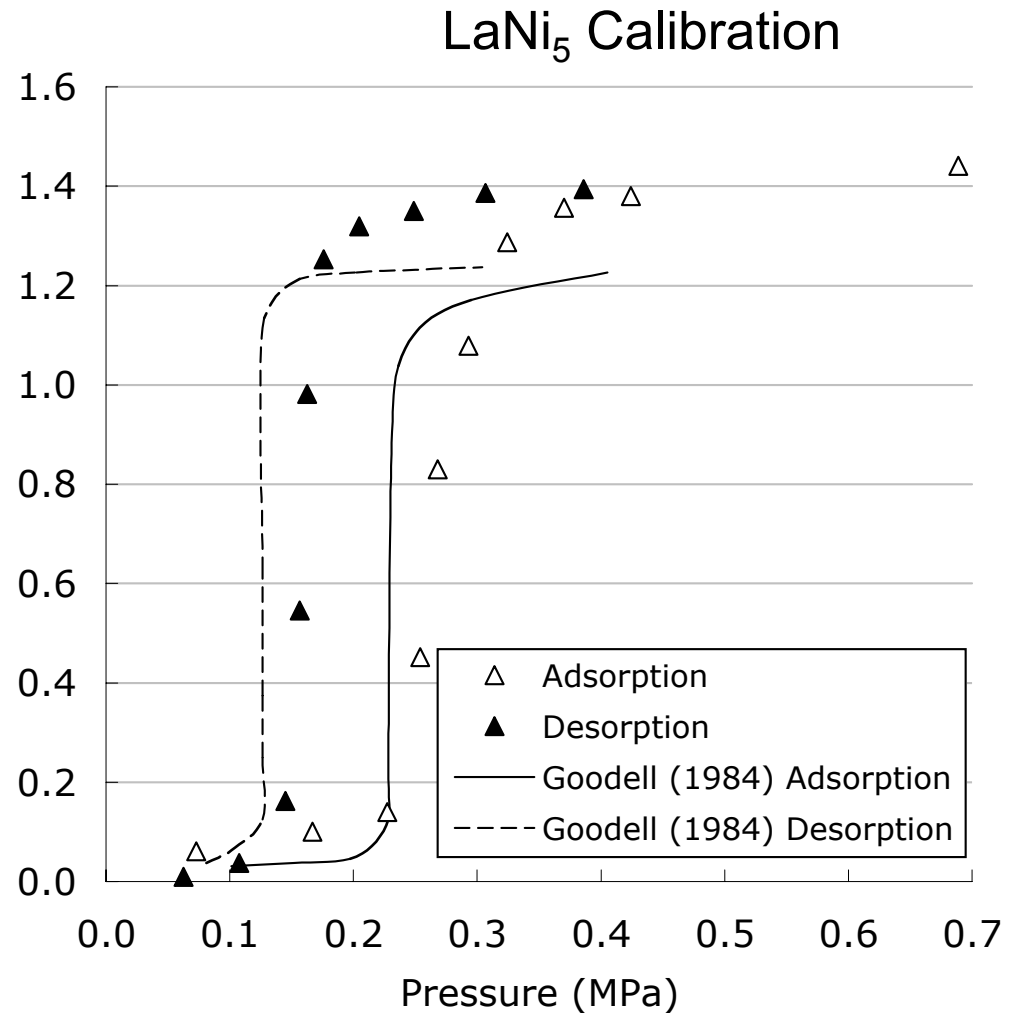
Source: AX-21 Bridge Data: Unpublished Work, Lachawiec, Qi & Yang (2005)



# High Pressure Measurement



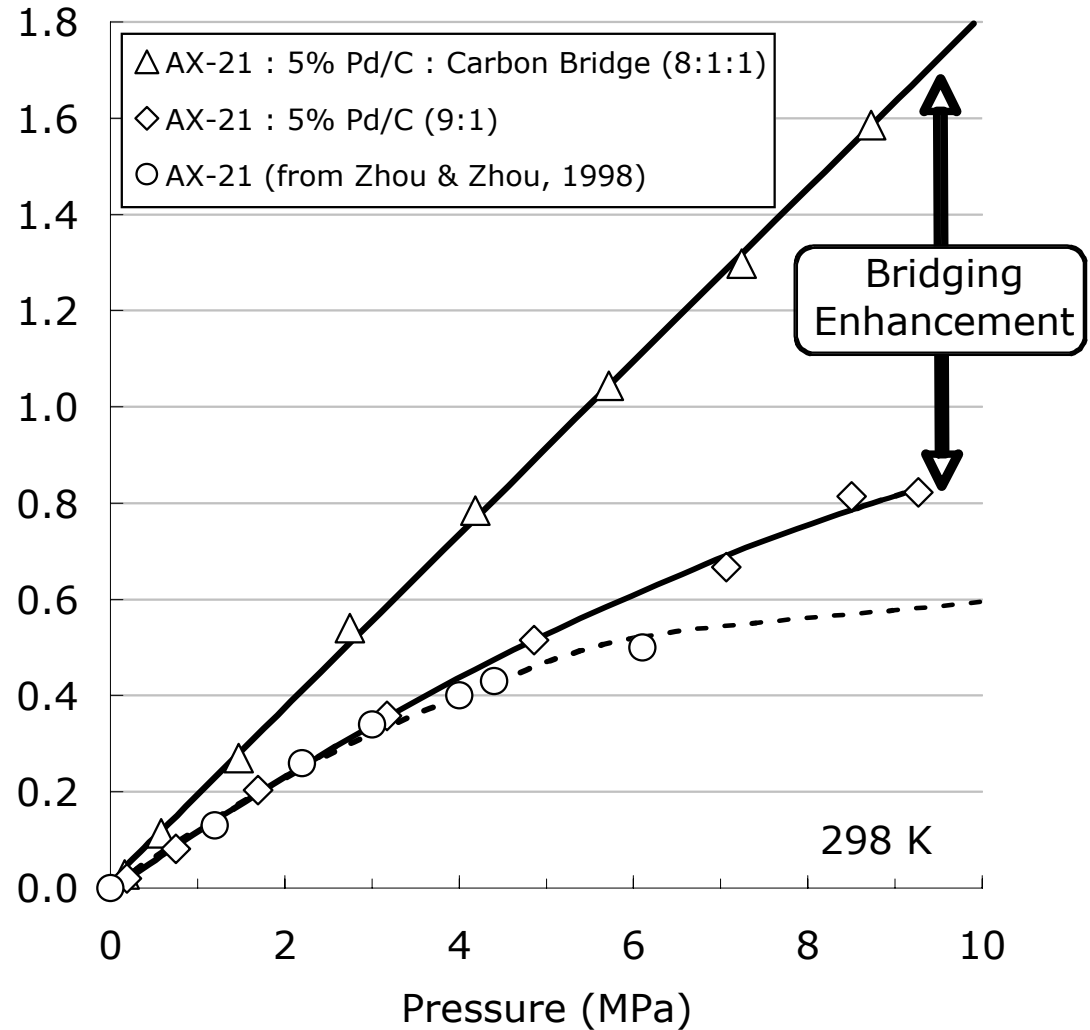
- Calibrated Volumetric System ( $\text{LaNi}_5$  &  $\text{TiAl}_{0.12}\text{V}_{0.04}$ )
- In-situ Pretreatment to 1023 K (750 C)
- Adsorption Measurements to 12 MPa (1800 psia)



Source: Goodell (1984) *J. Less-Common Met.* 99, 1

# High Pressure Spillover Enhancement

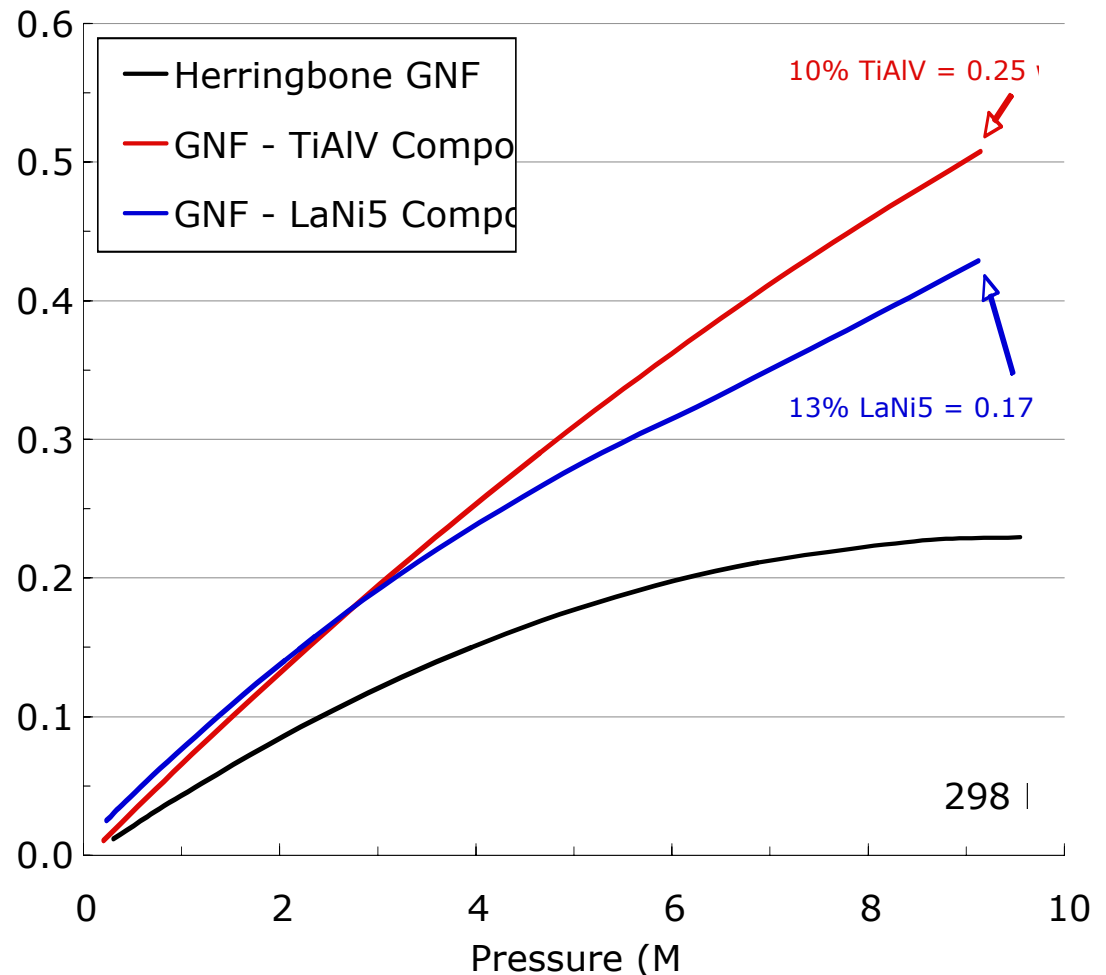
- Extension of low-pressure work
- Identical trends observed at high pressure
- Completely reversible adsorption at 298 K
- Adsorption capacity tripled at 10 MPa (only 1.3 times without bridges)
- 1.8 wt% capacity at 10 MPa without optimization



Sources: AX-21 Bridge Data: Unpublished Work, Lachawiec, Qi & Yang (2005)  
AX-21 Isotherm: Zhou & Zhou (1998) *Chem. Eng. Sci.* 53, 2531

# Graphite Nanofiber- Metal Hydride Composites

- Herringbone GNF
- $\text{LaNi}_5$  &  $\text{TiAl}_{0.12}\text{V}_{0.04}$  alloy powders 50 - 500 micron
- Incipient wetness-paste direct doping technique
- Isotherms are sum of individual component capacities
- No spillover at 298 K from these metal alloys
- Investigating catalysts & conditions to promote spillover



Source: Unpublished Work, Lachawiec and Yang (2005)

# Future Work

- Remainder of FY 2005 (3.5 wt%)
  - Screen & select optimal GNF synthesis catalyst
  - Extend demonstrated bridge-building technique to GNF composite materials
  - Develop mechanistic understanding for spillover
- FY 2006 (4.5 wt%)
  - Optimize catalyst/pretreatment conditions
  - Optimize bridge-building conditions
  - Implement spillover to achieve target for storage

# Addenda

- Slides after this page will not be on the poster, but will be part of the Annual Review CD material.

# Hydrogen Safety

## Primary Hydrogen Hazard

- Potential energy release & fire from hydrogen leaking at high pressure from measurement system

## Secondary Hazards

- Issues related to the nature of hydrogen as a compressed gas (e.g. stored potential energy, asphyxiation)

# Hydrogen Safety

## Mitigation of Primary Hazard

- Use of high-integrity VCR<sup>®</sup> fittings in high pressure system (leak rate  $< 10^{-8}$  atm-cc/sec)
- Component pressure rating to 2000 psia
- Helium leak check at 1500 psia prior to hydrogen introduction
- Procedural control to evacuate sample for minimum 1 hr to remove air from system
- Backfill with helium prior to sample removal
- System volume  $< 50$  cc & isolated during static measurements to limit leakage quantities